

PURIFICATION AND PROPERTIES OF THE  $H_2$ -OXIDIZING (UPTAKE) HYDROGENASE OF  
THE  $N_2$ -FIXING ANAEROBE *CLOSTRIDIUM PASTEURIANUM* W5

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**SUMMARY:** *Clostridium pasteurianum* has two distinct hydrogenases, the bidirectional hydrogenase and the  $H_2$ -oxidizing (uptake) hydrogenase. The  $H_2$ -oxidizing hydrogenase has been purified (up to 970-fold) to a specific activity of 17,600  $\mu\text{mol } H_2$  oxidized/min-mg protein (5 mM methylene blue) or 3.5  $\mu\text{mol } H_2$  produced/min-mg protein (1 mM methyl viologen). The uptake hydrogenase has a  $M_r$  of 53,000 (one polypeptide chain). Depending upon how protein was measured, the Fe and S<sup>=</sup> contents (gatom/mol) were 4.7 and 5.2 (by the dye-binding assay) or 7.2 and 8.0 (by the Lowry method). Both reduced and oxidized forms of the enzyme gave electron paramagnetic resonance signals. The activation energy for  $H_2$ -production and  $H_2$ -oxidation by the uptake hydrogenase was 59.1 and 31.2 kJ/mol, respectively. In the exponential phase of growth, the ratio of uptake hydrogenase/bidirectional hydrogenase in  $NH_3$ -grown cells was much lower than that in  $N_2$ -fixing cells.

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In 1978, we reported the discovery of a second hydrogenase, the  $H_2$ -oxidizing (uptake) hydrogenase, in the  $N_2$ -fixing anaerobe *Clostridium pasteurianum* W5 (1). The presence of this enzyme has since been confirmed by two other laboratories (2; M. W. W. Adams and L. E. Mortenson, personal communication). *C. pasteurianum* has a bidirectional hydrogenase (ferredoxin: $H^+$  oxidoreductase EC 1.18.3.1) which is involved in  $H_2$  production and it has been extensively studied (3-9). We have purified the uptake  $H_2$ ase and compared the properties of the two  $H_2$ ases. Based upon their different catalytic and structural properties, we conclude that the two  $H_2$ ases are chemically distinct. Although both  $H_2$ ases are found in  $N_2$ -fixing and  $NH_3$ -grown *C. pasteurianum* cells, their relative levels differ under these two growth conditions. The uptake  $H_2$ ase perhaps plays a role in regulating the redox state of ferredoxin which is the electron donor for nitrogenase; however the exact role of this enzyme remains to be established.

#### MATERIALS AND METHODS

All conditions, materials and methods were the same as in reference (1) except for the following:

**Growth of organism.** For enzyme purification,  $N_2$ -fixing cultures were grown in 40-L batches with the pH maintained at 5.5 with KOH. The cultures were harvested when  $A_{550\text{ nm}}$  reached 3-4

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**Abbreviations:**  $H_2$ ase, hydrogenase; Fd, ferredoxin; Fld, flavodoxin; MB, methylene blue; MV, methyl viologen; BV, benzyl viologen.

(Spectronic 20); the cells were in late exponential or between exponential and stationary phases. The cells were stored in liquid  $N_2$  in more recent studies, but were first frozen in liquid  $N_2$  and then stored at  $-20^\circ C$  under  $H_2$  in earlier studies. The activity of both  $H_2$ ases decreased with time at  $-20^\circ C$ , especially when cells were under aerobic conditions. When studying enzyme levels in  $N_2$ -fixing and  $NH_3$ -grown cells, the cultures were grown in medium (8L) containing 5 g/L  $CaCO_3$  for pH control. The cultures were sampled when  $A_{550\text{ nm}}$  was about 1, 2, 3, and 4.

**Preparation of crude extract.** Frozen cells were thawed at room temperature and under  $H_2$  in Tris-Cl (pH 8.5, 50 mM) containing lysozyme (1mg/mL) and DNase (0.1 mg/mL) for 2 hrs. Enzyme purification started with 0.5 or 1 kg of cell paste which was thawed at 1 g cells per 2 mL of buffer. Crude extract was the supernatant after centrifugation at  $35,000 \times g$  for 20 min. All operations were carried out under a  $H_2$  or  $N_2$  atmosphere and were at room temperature unless otherwise specified. All anaerobic buffers contained 1 mM of  $Na_2S_2O_4$ .

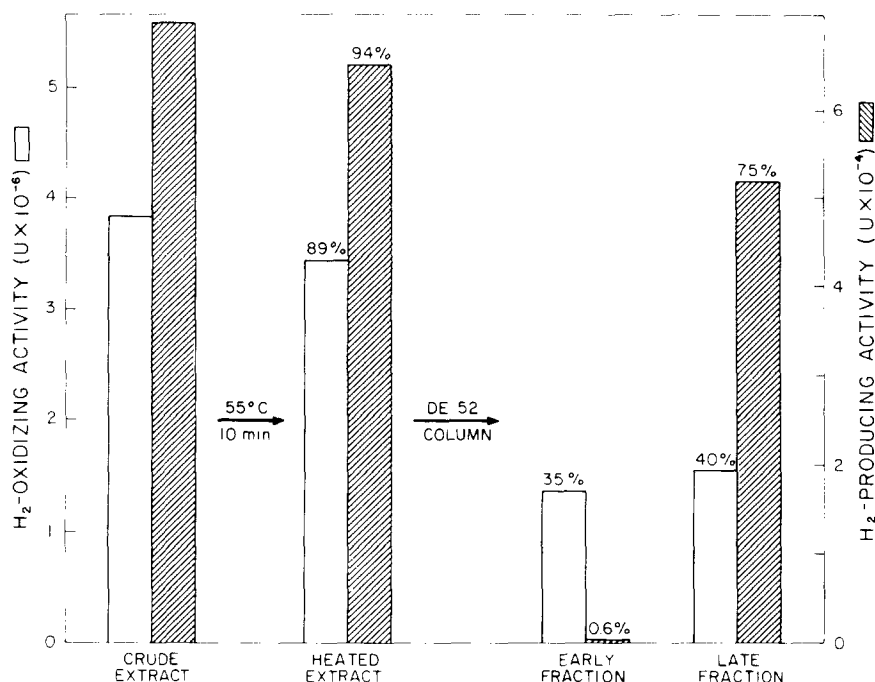
**Enzyme assays.** The  $H_2$ -oxidation assay was carried out with appropriate amounts of hydrogenase so that the rate of  $H_2$  uptake was below  $30 \mu\text{L}/\text{min}$ . One unit is the oxidation or production of  $1 \mu\text{mol } H_2/\text{min}$ . The  $H_2$  uptake activities reported in (1) were greatly underestimated because too much enzyme was used in the assay. The activation energy was measured in potassium phosphate buffer (pH 8, 50 mM).

**Analytical methods.** Iron (3) and inorganic sulfide (10) were measured as described. Core extrusion was by the method of Gillum et al. (6). Electron paramagnetic resonance was measured with a Varian E9 spectrometer. Sodium dodecyl sulfate (SDS)-gel electrophoresis was performed according to (11) and (12). Peptide mapping (13) was performed as described (14). Unless specified otherwise, protein was measured by the dye-binding method (15) as described by BioRad Technical Bulletin No. 1051. For the uptake  $H_2$ ase, the dye-binding assay gave a protein concentration 55% higher than that given by the Lowry method (16). Thus, a sample with a protein concentration of 5.51 mg/mL (dye-binding assay) became 3.57 mg/mL (Lowry method).

**Enzyme purification.** Heat treatment of crude extract was described (3). The first DE 52 column step was as in (1); a column of  $5 \times 15\text{ cm}$  was used for the extract from 0.5 kg of cells. Using this column/extract ratio, the uptake  $H_2$ ase was not adsorbed to DE 52 and the effluent containing the uptake  $H_2$ ase was designated the early fraction. The bidirectional  $H_2$ ase was eluted at about 0.17 M chloride, which was designated the late fraction. The first hydroxyapatite (BioRad HTP) column was  $5 \times 12\text{ cm}$ , and was eluted by a linear gradient of potassium phosphate (0.01-0.4 M, pH 7.2, total volume 800 ml, also containing 0.1 M KCl).  $H_2$ ase was eluted between 90 and 220 ml, and it was concentrated by ultrafiltration with an Amicon PM-10 membrane before being applied to the next column. The Sephadex G-100 column ( $5 \times 56\text{ cm}$ ) was eluted as in (1).  $H_2$ ase-containing fractions were diluted with an equal volume of 10 mM Tris-Cl (pH 8) and then loaded onto the second DE 52 column. This column ( $2.5 \times 19.5\text{ cm}$ ) was eluted by a linear gradient of KCl (0.01-0.2 M, total volume 600 ml) in Tris-Cl (50 mM, pH 8).  $H_2$ ase was eluted between 355 and 395 ml. The final HTP column ( $2.5 \times 12\text{ cm}$ ) was eluted by a linear gradient of potassium phosphate (0.04-0.15 M, pH 7.2, total volume 400 ml, also containing 0.1 M KCl).  $H_2$ ase was eluted between 117 and 142 ml as a yellow-brown band. Between column steps,  $H_2$ ase samples were stored at  $5-10^\circ C$ . The purified enzyme was stored in liquid  $N_2$  as frozen pellets.

## RESULTS AND DISCUSSION

The two  $H_2$ ases of *C. pasteurianum* were easily separated by DEAE-cellulose (DE 52) column chromatography. Fig. 1 shows that the  $H_2$ -oxidizing and  $H_2$ -producing activities of a crude extract were nearly quantitatively recovered after the heat treatment, with the ratio of  $H_2$ -oxidation/ $H_2$ -production remaining at 54-56. After the first DE 52 column step, the  $H_2$ -oxidizing (uptake) activity was roughly evenly distributed between the early and the late fractions, whereas the  $H_2$ -producing activity eluted essentially in the late fraction. The late fraction corresponds to the previously characterized bidirectional  $H_2$ ase, and its  $H_2$ -oxidation/ $H_2$ -production activity ratio was 30 at this point (the ratio for purified bidirectional  $H_2$ ase is 25). The early fraction had a  $H_2$ -oxidation/ $H_2$ -production activity ratio of 3,500, and the  $H_2$ ase in this fraction is designated the  $H_2$ -oxidizing (uptake)  $H_2$ ase. Further purified uptake  $H_2$ ase had a  $H_2$ -oxidation/ $H_2$ -production



**Figure 1.** Separation of the H<sub>2</sub>-oxidizing H<sub>2</sub>ase (the early fraction) and the bidirectional H<sub>2</sub>ase (the late fraction) of *C. pasteurianum* by a DEAE-cellulose (DE 52) column. An extract from 1 kg of cells was used, and conditions were given in Materials and Methods. Percentages denote the recovery of activities by taking the total activity in the crude extract as 100%.

activity ratio of 5,000; the H<sub>2</sub>-producing activity of the uptake H<sub>2</sub>ase is negligible for practical purposes. Using the H<sub>2</sub>-oxidation/H<sub>2</sub>-production activity ratio of the bidirectional H<sub>2</sub>ase and the H<sub>2</sub>-oxidizing and H<sub>2</sub>-producing activities of a crude extract, we can deduce the amount of the two H<sub>2</sub>ases, in terms of H<sub>2</sub>-uptake activities, in crude extracts. In crude extracts of N<sub>2</sub>-fixing cells (with the H<sub>2</sub>-oxidation/H<sub>2</sub>-production activity ratio near 55), about 50% of the H<sub>2</sub>-uptake activity was from the uptake H<sub>2</sub>ase.

Table 1 shows the results of one preparation of the uptake H<sub>2</sub>ase from 0.5 kg of cells. The peak fraction had an activity of 22,300 units/mg. Another preparation from 1 kg of cells yielded a final activity of 15,540 units/mg. We also tried to replace the first HTP column step with (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> fractionation (H<sub>2</sub>ase was in the supernatant of 60% saturated (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>) and an Octyl-Sepharose column step (eluted by a gradient of 60-0% saturated (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> in 50 mM Tris-Cl, pH 8, at 10°C). The latter procedure gave a final activity of 15,900 units/mg (peak = 17,600 units/mg) with a yield of 6%. Because the starting activity varied from 17.6 to 25 units/mg, the fold of purification was between 700 and 970. The purity of the enzyme was examined by SDS-gel electrophoresis followed by densitometry. The intensity of the H<sub>2</sub>ase band (M<sub>r</sub> = 53,000) was proportional to activity in

TABLE 1. Purification of the H<sub>2</sub>-Oxidizing (Uptake) Hydrogenase from *C. pasteurianum*

Step	Volume (ml)	Total protein (mg)	Spec. Act. (units/mg protein)	Fold of purification	Total activity (units)	Recovery of Activity (%)
Crude extract	900	27,540	17.6 <sup>a</sup>	1	484,700	100
Heat treatment	850	10,900	45.2 <sup>a</sup>	2.6	492,680	102
First DE 52 column	700	2,933	128	7.3	375,420	77
First HTP column	131	704	443 (peak = 687)	25	311,870	64
Sephadex G-100 column	180	274	923 (peak = 1,374)	52	252,900	52
Second DE 52 column	40	45	4,716 (peak = 6,294)	268	212,220	44
Second HTP column	25	4.05	17,020 (peak = 22,300)	967	68,930	14

<sup>a</sup>The total H<sub>2</sub>-uptake activities for these samples were 30.6 and 75.2 units/mg, respectively, whereas the H<sub>2</sub>-producing activities were 0.52 and 1.2 units/mg, respectively. The contribution of H<sub>2</sub>-uptake activity by the bidirectional H<sub>2</sub>ase was subtracted by using a H<sub>2</sub>-uptake/H<sub>2</sub>-production activity ratio of 25 for the enzyme. There was negligible H<sub>2</sub>-producing activity after the fourth step.

the sample. For a sample with an activity of 15,900 units/mg, the H<sub>2</sub>ase band represented 84.6% of the Coomassie blue-stained material.

Table 2 lists the catalytic and structural properties of the uptake H<sub>2</sub>ase and a comparison of the two H<sub>2</sub>ases of *C. pasteurianum*. For the uptake H<sub>2</sub>ase, a M<sub>r</sub> of 53,000 was obtained with both electrophoretic methods (11, 12), but gel filtration on a Sephacryl S-200 column gave a M<sub>r</sub> = 48,500. We used the M<sub>r</sub> of 53,000 for calculations. Depending upon the methods of protein determination (see MATERIALS AND METHODS), the activity (units/mg) and the Fe/S= contents (gatom/mol) of a sample can be 15,540, 4.7, and 5.2, respectively (using the dye-binding protein assay), or 23,990, 7.2, and 8.0 (using the Lowry protein assay). We do not yet know which protein assay is more accurate for this enzyme. Core extrusion of a sample (15,540 units/mg) quantitatively yielded FeS clusters of the 4Fe-4S type. The H<sub>2</sub>/dithionite-reduced enzyme (15,900 units/mg) had a yellow-brown color and had broad absorption in the 400 nm region, with an  $\epsilon$  (400 nm) = 12.8 mM<sup>-1</sup> cm<sup>-1</sup> (dye-binding assay) or 19.8 mM<sup>-1</sup> cm<sup>-1</sup> (Lowry assay). More concentrated samples (5.5 mg/ml) showed a greenish brown color under N<sub>2</sub>. Fig. 2 shows the electron paramagnetic resonance spectra of the enzyme. Judging from the intensity of the signals near g=2 in (b) and the ease to generate the oxidized form, we speculate that the minor signal near g=2.0 in (a) was from the oxidized enzyme. Additional redox titrations will be needed to clarify this point and to allow interpretation of these signals.

The significantly higher E<sub>a</sub> for H<sub>2</sub>-production by the uptake H<sub>2</sub>ase (Table 2) may explain the enzyme's low capacity to produce H<sub>2</sub>. Fig. 3 shows the temperature dependence of H<sub>2</sub> production

TABLE 2. Properties of H<sub>2</sub>-Oxidizing H<sub>2</sub>ase and Bidirectional H<sub>2</sub>ase of *C. pasteurianum*

	H <sub>2</sub> -Oxidizing H <sub>2</sub> ase	Bidirectional H <sub>2</sub> ase <sup>a</sup>
Molecular Weight	53,000	60,000
Polypeptide	1	1
Iron (gatom/mol)	4.7 <sup>b</sup> or 7.2 <sup>c</sup>	12 <sup>c</sup>
Acid-labile sulfide (gatom/mol)	5.2 <sup>b</sup> or 8.0 <sup>c</sup>	12 <sup>c</sup>
Structure of FeS center	n x (4 Fe-4S)	3 x (4 Fe-4S)
Catalytic activities (μmol H <sub>2</sub> /min-mg)		
H <sub>2</sub> -oxidation		
Spec. Act. (5 mM MB)	15,000 <sup>b</sup>	13,700 <sup>c</sup>
Km for MB (mM)	0.42	ND
Km for Fd (μM)	42.2	ND
Km for H <sub>2</sub> (atm)	0.19	0.57 <sup>d</sup>
H <sub>2</sub> -production		
Spec. Act. (1 mM MV)	3 <sup>b</sup>	550 <sup>c</sup>
Km for MV (mM)	0.36	6.2
Km for Fd (μM)	ND	51
H <sub>2</sub> -oxidation/H <sub>2</sub> -production	5,000	25
Activation energy (kJ/mol)		
H <sub>2</sub> -oxidation	31.2	22.9 <sup>d</sup>
H <sub>2</sub> -production	59.1	22.7 <sup>d</sup>
Electron carriers	Fd, Fld, MV BV, MB, FAD, FMN, riboflavin.	Fd, Fld, MV BV, MB, FAD <sup>d</sup> , FMN <sup>d</sup> , riboflavin <sup>d</sup> .

<sup>a</sup>From references (3, 6).<sup>b</sup>Protein determined by the dye-binding assay.<sup>c</sup>Protein determined by the Lowry method.<sup>d</sup>This study.

ND, not determined.

by the uptake H<sub>2</sub>ase. Fig. 4 shows the effect of pH on activities of the uptake H<sub>2</sub>ase. At pH 8, Tris and phosphate buffers gave similar activities, but the bidirectional H<sub>2</sub>ase is twice as active in phosphate than in Tris buffer at pH 8 (5). Between pH 8 and 9, there was an abrupt increase in H<sub>2</sub>-producing activity by the uptake H<sub>2</sub>ase; however, the intracellular pH is not likely to be in this range (17). The effects of temperature and pH may provide a way of probing the mechanism of H<sub>2</sub>-formation by this enzyme.

We also compared the primary structure of the two H<sub>2</sub>ases of *C. pasteurianum* by the peptide mapping method (13, 14). Fig. 5 shows that the two H<sub>2</sub>ases do not share a common polypeptide. Therefore, we conclude that the two H<sub>2</sub>ases are chemically distinct.

We studied extensively the levels of the two H<sub>2</sub>ases in N<sub>2</sub>-fixing, NH<sub>3</sub>-grown, NH<sub>3</sub> ⇌ N<sub>2</sub> transitional, and energetically perturbed cultures of *C. pasteurianum*. When we compared the level of the two H<sub>2</sub>ases in cells grown under different conditions and harvested at different stages of growth we observed much variation in the absolute level of the two H<sub>2</sub>ases (e.g., Table 3). However, when exponentially growing cells were compared at similar culture densities (A<sub>550 nm</sub> = 1-2), the N<sub>2</sub>-

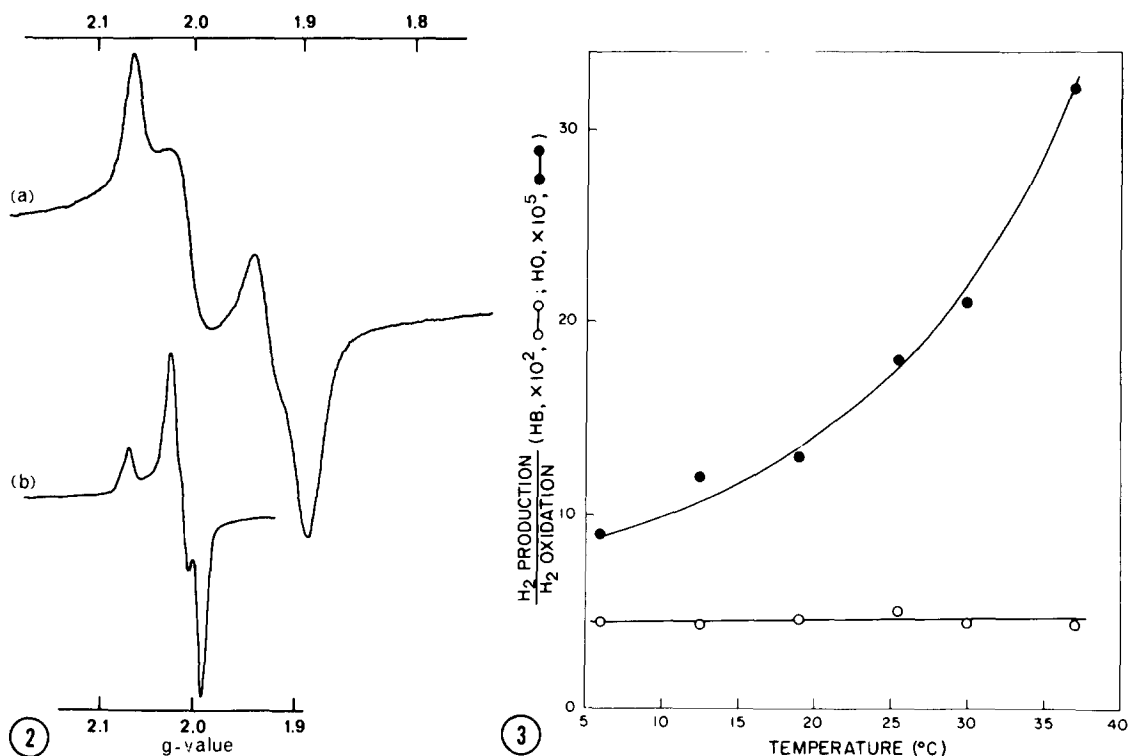


Figure 2. Electron paramagnetic resonance spectra of the uptake  $H_2$ ase of *C. pasteurianum*. (a) The  $H_2$ ase sample (2.3 mg/ml, 15,750 units/mg) was as isolated in potassium phosphate buffer (pH 7.2, about 0.1 M; with 0.1 M KCl). EPR conditions were: Microwave frequency, 9.260 GHz; temperature, 13 K; power, 50 mW; modulation, 4 mT; gain,  $2 \times 10^3$ . (b) The  $H_2$ ase sample (15,540 units/mg) was thawed and diluted with half volume of dist.  $H_2O$  under  $N_2$  to give a protein concentration of 3.67 mg/ml. EPR conditions were: microwave frequency, 9.255 GHz; temperature, 16 K; power, 20 mW; modulation 2.5 mT; gain,  $1.6 \times 10^3$ . (The signal height in (b) was reduced by 50% during graph preparation to bring the g-values to scale.)

Figure 3. Effect of temperature on the relative rates of  $H_2$ -production and  $H_2$ -oxidation by the uptake  $H_2$ ase ( $\bullet$ ) and the bidirectional  $H_2$ ase ( $\circ$ ) of *C. pasteurianum*. The assays were performed in 50 mM potassium phosphate buffer, pH 8, using manometry. The volume of gas was corrected for each temperature.

fixing cells always had a lower level of the bidirectional  $H_2$ ase and a higher ratio of uptake  $H_2$ ase/bidirectional  $H_2$ ase than  $NH_3$ -grown cells (Table 3). Because the bidirectional  $H_2$ ase can lower the level of reduced Fd in the cell whereas the uptake  $H_2$ ase may elevate it, the relative level of the two  $H_2$ ases may have an effect on the level of reduced Fd or the electron-donating potential of Fd in the cell. Since reduced Fd is the electron donor for nitrogenase (reduced Fd:dinitrogen oxidoreductase EC 1.18.2.1), the bidirectional  $H_2$ ase and nitrogenase compete against each other for reduced Fd, and the bidirectional  $H_2$ ase may also lower the ratio of reduced Fd/oxidized Fd to hamper nitrogenase activity. The redox potential required for half-maximal nitrogenase activity has been reported to be between -460 and -470 mV (18-20); Braaksma et al. (22) reported that nitrogenase activity begins to decrease when the potential is above -440 mV. The bidirectional  $H_2$ ase of

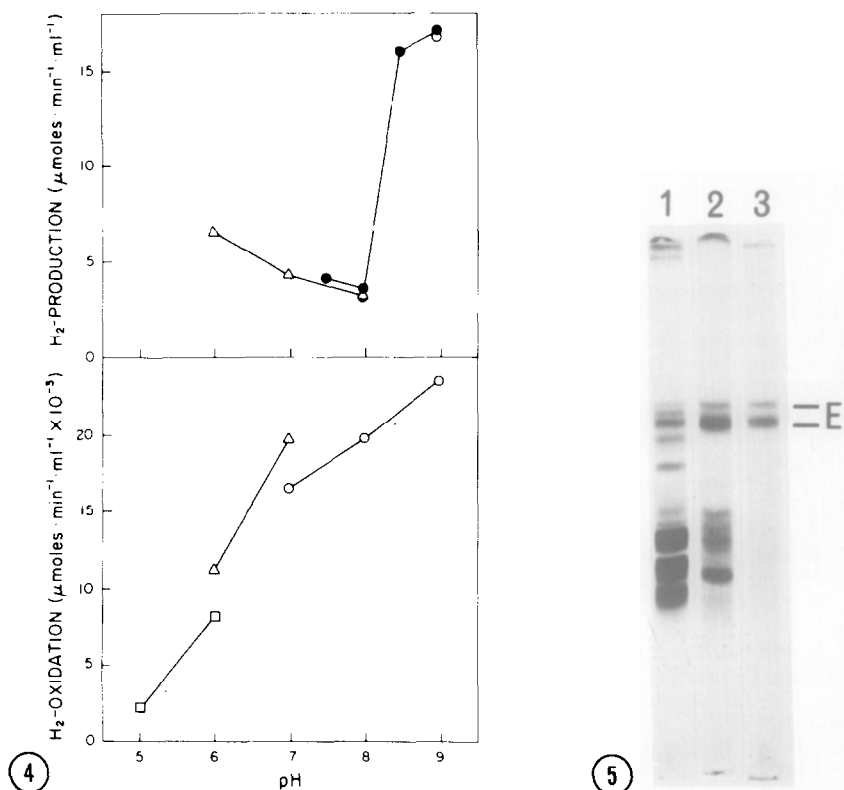


Figure 4. Effect of pH on the activities of the uptake H<sub>2</sub>ase of *C. pasteurianum*. The assays were carried out in 50 mM of sodium acetate (□, pH 5-6), potassium phosphate (Δ, pH 6-8), and Tris-Cl (○, ●, pH 7-9) buffers. Methylene blue (5 mM) and methyl viologen (1 mM) were used as the electron acceptor and donor.

Figure 5. Peptide mapping following limited proteolysis of the uptake H<sub>2</sub>ase (1) and the bidirectional H<sub>2</sub>ase (2) of *C. pasteurianum*. Gel 3 contained proteinase only, with E marking the bands of the proteinase (*S. aureus* V8 proteinase). The activities of the samples were: uptake H<sub>2</sub>ase, 15,540 units/mg; bidirectional H<sub>2</sub>ase, 420 units/mg (H<sub>2</sub>-production).

*C. pasteurianum* can produce H<sub>2</sub> when the potential is more positive than these values (8). We speculate that the uptake H<sub>2</sub>ase facilitates N<sub>2</sub>-fixation in *C. pasteurianum* by maintaining a more favorable level of reduced Fd and/or a more favorable ratio of reduced Fd/oxidized Fd for nitrogenase. The speculation remains to be tested.

TABLE 3. Levels of H<sub>2</sub>-Oxidizing H<sub>2</sub>ase and Bidirectional H<sub>2</sub>ase in Crude Extracts of N<sub>2</sub>-Fixing and NH<sub>3</sub>-Grown Cells of *C. pasteurianum*.

Growth Phase	Growth Conditions	H <sub>2</sub> ase Level (units/mg protein) <sup>a</sup>		HO/HB
		H <sub>2</sub> -oxidizing (HO)	Bidirectional (HB)	
Exponential (A <sub>550 nm</sub> ≈ 2)	N <sub>2</sub> -fixing NH <sub>3</sub> -grown	39	35	1.1
		30	75	0.4
Stationary (A <sub>550 nm</sub> ≈ 4)	N <sub>2</sub> -fixing NH <sub>3</sub> -grown	15	14	1.1
		46	36	1.3

<sup>a</sup>The H<sub>2</sub>ase level is based upon the H<sub>2</sub>-uptake activity of each enzyme. See footnote to Table 1 for methods of calculation.

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## REFERENCES

1. Chen, J. S., and Blanchard, D. K. (1978) *Biochem. Biophys. Res. Commun.* 84, 1144-1150.
2. Schneider, K., Pinkwart, M., and Jochim, K. (1983) *Biochem. J.* 213, 391-398.
3. Chen, J. S., and Mortenson, L. E. (1974) *Biochim. Biophys. Acta* 371, 283-298.
4. Chen, J. S., Mortenson, L. E., and Palmer, G. (1976) in *Iron and Copper Proteins*, Yasunobu, K. T., Mower, H. F., and Hayaishi, O., Eds., pp. 68-82, Plenum Press, New York.
5. Mortenson, L. E., and Chen, J. S. (1976) in *Microbial Production and Utilization of Gases ( $H_2$ ,  $CH_4$ ,  $CO$ )*, Schlegel, H. G., Gottschalk, G., and Pfennig, N., Eds., pp. 97-108, E. Goltze KG, Gottingen.
6. Gillum, W. O., Mortenson, L. E., Chen, J. S., and Holm, R. H. (1977) *J. Am. Chem. Soc.* 99, 584-595.
7. Erbes, D. L., and Burris, R. H. (1978) *Biochim. Biophys. Acta* 525, 45-54.
8. Chen, J. S. (1978) in *Hydrogenases: Their Catalytic Activity, Structure and Function*, Schlegel, H. G., and Schneider, K., Eds., pp. 57-80, E. Goltze KG, Gottingen.
9. Adams, M. W. W., Mortenson, L. E., and Chen, J. S. (1980) *Biochim. Biophys. Acta* 594, 105-176.
10. Chen, J. S., and Mortenson, L. E. (1977) *Anal. Biochem.* 79, 157-165.
11. Weber, K., and Osborn, M. (1975) in *The Proteins*, Neurath, H., and Hill, R. L., Eds., 3rd Edn., Vol. 1, pp. 179-223, Academic Press, New York.
12. Laemmli, U. K. (1970) *Nature* 227, 680-685.
13. Cleveland, D. W., Fisher, S. G., Kirschner, M. W., and Laemmli, U. K. (1977) *J. Biol. Chem.* 252, 1102-1106.
14. Jin, S.-L. C., Blanchard, D. K., and Chen, J. S. (1983) *Biochim. Biophys. Acta* 748, 8-20.
15. Bradford, M. M. (1976) *Anal. Biochem.* 73, 248-254.
16. Lowry, O. H., Rosebrough, N. J., Farr, A. L., and Randall, R. J. (1951). *J. Biol. Chem.* 143, 265-275.
17. Riebeling, V., Thauer, R. K., and Jungermann, K. (1975) *Eur. J. Biochem.* 55, 445-453.
18. Evans, M. C. W., and Albrecht, S. L. (1974) *Biochem. Biophys. Res. Commun.* 61, 1187-1192.
19. Hallenbeck, P. C. (1983) *Arch. Biochem. Biophys.* 220, 657-660.
20. Lough, S., Burns, A., and Watt, G. D. (1983) *Biochemistry* 22, 4062-4066.
21. Braaksma, A., Haaker, H., Grande, H. J., and Veeger, C. (1982) *Eur. J. Biochem.* 121, 483-491.